

Name: _____

Period: _____

Properties of Water Lab with Stats

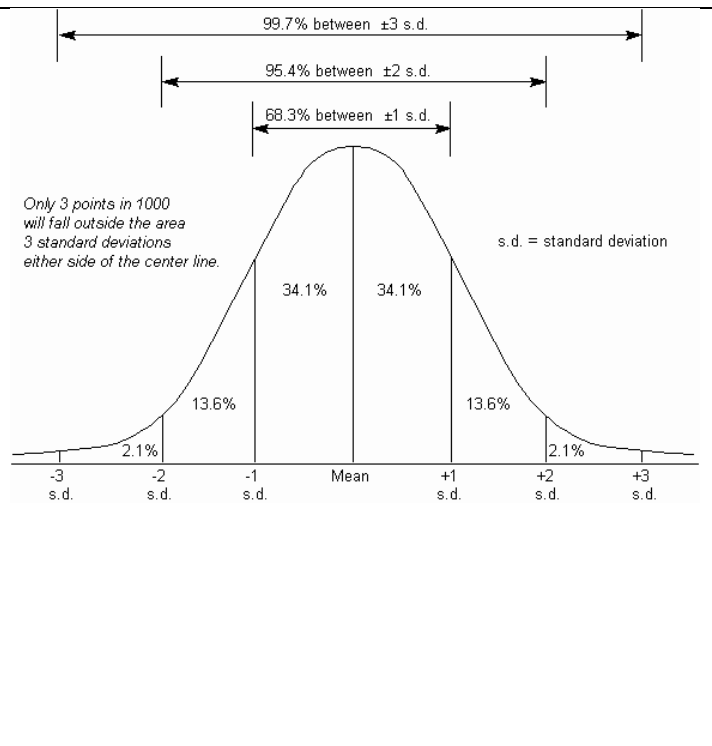
(modified from a lab created by Crystal Jenkins Stawieri)

Part I. Standard Deviation and Standard Error

When you measure something (like the number of drops of water that can stay on a penny without overflowing), how do you know whether that measurement is valid? Is it representative of *all* pennies? All droppers? All water?

One way to improve the quality of collected data is to increase the sample size, calculate a mean (an average), and then to determine the **standard deviation**. Standard deviation (often reported as “+/-”) shows how much variation there is from the mean. When data points are *close* together, the standard deviation will be *small*. If data points are *spread out*, the standard deviation will be *larger*.

Typical data will show a **normal distribution** (a bell-shaped curve). In a normal distribution, about 68% of values are within one standard deviation of the mean, 95% of values are within two standard deviations of the mean, and 99% of the values are within three standard deviations of the mean. The formula for standard deviation is shown below, where \bar{x} is the mean, x_i is any given data value, and n is the sample size. Consider the following sample problem.



Grades on the most recent AP Biology quiz were as follows: 96, 96, 93, 90, 88, 86, 86, 84, 80, 70.

- Step 1:** Find the **Mean** (). _____
- Step 2:** Determine the **Deviation** ($x_i - \bar{x}$)² from the mean for each value and square it, then add up all of the total values. _____
- Step 3:** Calculate the **Degrees of Freedom (n-1)**. _____
- Step 4:** Put it all together to find **s**. _____

Standard Deviation

$$s = \sqrt{\frac{\sum (x_i - \bar{x})^2}{n - 1}}$$

If you're like me, you're going to want to organize this into a table. If you can, try to solve what's below without looking at the answers on the bottom (or, at least, try to do that as much as possible).

Scores	96	96	93	90	88	86	86	84	80	70	Total	Average
$x_i - \bar{x}$												
$(x_i - \bar{x})^2$												

$\sum (x_i - \bar{x})^2 =$ _____	$n - 1 =$ _____	$\sum (x_i - \bar{x})^2 / n - 1 =$ _____	$s =$ _____
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In problem above, here are the values you should be getting:

- the mean is 87
- The sum of $x_i - \bar{x}$ is 556.9
- The number of samples is 10, so $n - 1 = 9$
- $556.9/9 = 61.87$
- The square root of 61.87 is 7.8. Working with significant figures, that rounds to 8. That's our standard deviation.

So, in this sample, one standard deviation would be (87 - 8) through (87 + 8), or 79 to 95. That means that 68% of the data should fall between these numbers. Two standard deviations would be (87 - 16) through (87 + 16), or 71 to 103. At two standard deviations, 95% of the data should fall between these numbers. Three standard deviations would be (87 - 24) through (87 + 24), or 63 to 111. At three standard deviations, 99% of the data should fall between these numbers.

Standard error of the mean is used to represent our uncertainty in estimating the mean. It accounts for both sample size and variability. The formula used to calculate standard error of the mean is shown below. As standard error grows smaller, the likelihood that the sample mean is an accurate estimation of the population increases.

Using the data from the standard deviation example above, the mean is 87 and the standard deviation is 8. Plug in the numbers (remembering that *n* is 10).

Standard Error
$SE_{\bar{x}} = \frac{S}{\sqrt{n}}$

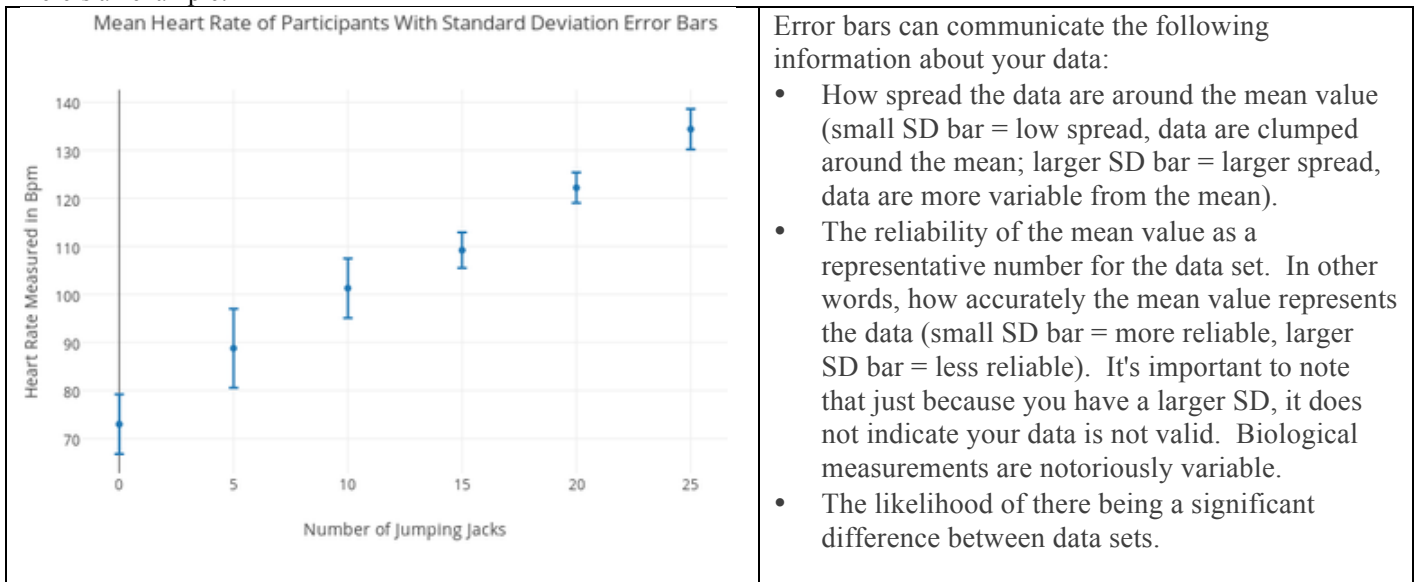
$$8/\sqrt{10} = 2.5$$

So, our standard error of the mean equals 2.5 This means that measurements vary by +/- 2.5 from the mean.

(The next section is slightly adapted from <https://www.biologyforlife.com/interpreting-error-bars.html>)

We use standard error of the mean to draw error bars. An error bar is a line through a point on a graph, parallel to one of the axes, which represents the uncertainty or variation of the corresponding coordinate of the point.

Here's an example:



A "significant difference" means that the results that are seen are most likely not due to chance or sampling error. In any experiment or observation that involves sampling from a population, there is always the possibility that an observed effect would have occurred due to sampling error alone. But if result is "significant," then the investigator may conclude that the observed effect actually reflects the characteristics of the population rather than just sampling error or chance.

The standard deviation error bars on a graph can be used to get a sense for whether or not a difference is significant. Look for overlap between the standard deviation bars:

When standard deviation errors bars overlap quite a bit, it's a clue that the difference is <i>not</i> statistically significant.	When standard deviation errors bars overlap even less, it's a clue that the difference is <i>probably not</i> statistically significant.	When standard deviation error bars do not overlap, it's a clue that the difference may be significant.

Standard Deviation Practice

Standard Deviation $s = \sqrt{\frac{\sum (x_i - \bar{x})^2}{n - 1}}$	Standard Error $SE_x = \frac{s}{\sqrt{n}}$	A group of students is measuring the number of stomata (a type of pore used to let in carbon dioxide and control water loss) / cm ² on the bottom surface of sunflower leaves. They took six measurements. Figure out the standard deviation and the standard error.

Sunflower Plant	1	2	3	4	5	6	TOTAL	Mean (\bar{x})	
Stomata (per cm ²)	88	93	90	92	75	78			
$x_i - \bar{x}$									
$(x_i - \bar{x})^2$									
$\sum (x_i - \bar{x})^2 =$ _____	$n-1 =$ ____.		$\sum (x_i - \bar{x})^2 / n-1 =$ _____				$s =$ _____	$SE_x =$ _____	

Part II: How does alcohol affect hydrogen bonds between different water molecules

Pre-Lab Questions: To stimulate your memory answer the following questions.

- Why is water considered to be polar? Draw a sketch of a water molecule to illustrate this polarity.
- Which type of bonds form between the oxygen and hydrogen atoms of **TWO DIFFERENT** water molecules? Draw a few water molecules, showing these bonds.

Some Background. We're going to measure hydrogen bond strength by counting the number of drops of water you can put on a dry penny. We'll do this twice: once with just water, and once with a combination of water and isopropyl alcohol. Isopropyl alcohol is very slightly polar, with its polarity caused by the hydroxyl group on the middle carbon.

So, form a hypothesis about how the results will differ and write it down below. Will adding a drop of alcohol increase water's ability to form hydrogen bonds, or decrease it? What will we see in the lab?.

$ \begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{C}-\text{C}-\text{H} \\ \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \\ \text{isopropyl alcohol} \end{array} $	My prediction:
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Materials: Penny, water, isopropyl alcohol (91%), pipettes, paper towel. **Safety:** Don't get the alcohol in your eyes.

Procedure:

- Obtain a DRY penny and place it on a DRY paper towel.
- Using a clean pipette, add distilled water to the penny drop by drop until it overflows. **Be sure to count the drops!** Record the number of drops for Trial 1 in Data Table 1 below.
- Repeat steps 1-2 for a total of five trials.
- Obtain a DRY penny and place it on a DRY paper towel.
- Using a pipette, place 10 drops of isopropyl alcohol on the penny.
- Using a clean pipette, add water to the penny drop by drop until it overflows. **Be sure to count the drops!** Record the number of drops for Trial 1 in Data Table 1 below.
- Repeat steps 4-6 for a total of five trials. **Be sure to add a new drop of alcohol between trials!**

Data Collection:

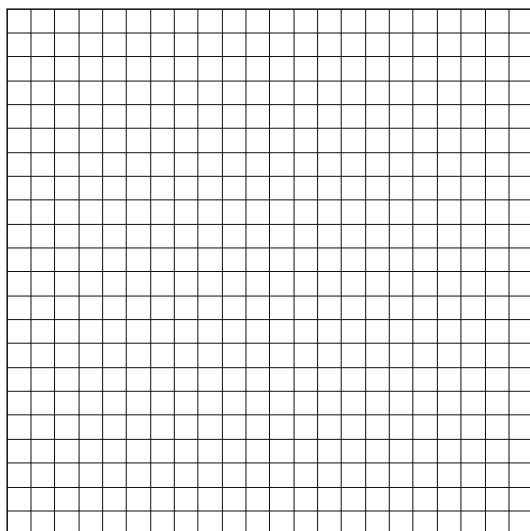
Data Table 1: Number of Drops of Held on the Surface of a Penny

	1	2	3	4	5	Total	Ave.		1	2	3	4	5	Total	Ave.
Water								Water+ alcohol							
$x_i - \bar{x}$															
$(x_i - \bar{x})^2$															

Water	$\Sigma (x_i - \bar{x})^2 =$ _____	$n-1 =$ _____	$\Sigma (x_i - \bar{x})^2 / n-1 =$ _____	$s =$ _____	$SE_{\bar{x}} =$ _____
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Water + alcohol	$\Sigma (x_i - \bar{x})^2 =$ _____	$n-1 =$ _____	$\Sigma (x_i - \bar{x})^2 / n-1 =$ _____	$s =$ _____	$SE_{\bar{x}} =$ _____
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Create an appropriately labeled bar graph to illustrate the sample means for the penny with and without alcohol. Add error bars indicating the standard error. (In other words, you'll have a bar graph and error bars) Include a title and labels for axes as needed.



C-E-R Conclusion

1. Make a **Claim** about how alcohol affects hydrogen bonds between water molecules.
2. Using data from this experiment, provide **Evidence** that supports the claim.
3. Using background knowledge and data from this lab, provide **Reasoning** that uses the evidence to justify the claim.